

Karima Horchani-Naifer and
Mokhtar Férid*Laboratoire des Matériaux, Institut National de
Recherche Scientifique et Technique, BP 95,
2050 Hammam-Lif, TunisiaCorrespondence e-mail:
mokhtar.ferid@inrst.rnrt.tn**Key indicators**Single-crystal X-ray study
 $T = 298\text{ K}$
Mean $\sigma(\text{P-O}) = 0.005\text{ \AA}$
 R factor = 0.035
 wR factor = 0.096
Data-to-parameter ratio = 19.4For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.**Potassium ytterbium diphosphate** KYbP_2O_7 consists of P_2O_7 double-tetrahedral groups connected to YbO_6 octahedra, forming a framework whose channels are occupied by K^+ cations.Received 21 October 2006
Accepted 17 December 2006**Comment**

Rare-earth containing compounds are widely used for optical applications such as colour displays, solid-state lasers, and sensors (Weber, 1999; Blasse & Grabmaier, 1994; Hashimoto *et al.*, 1991; Horchani-Naifer *et al.*, 2006). Yb^{3+} -doped crystals exhibit luminescence of Yb^{3+} from the charge-transfer state with emission in the near-UV–visible region and short radiative lifetimes (van Pieterse *et al.*, 2000; Guerassimova *et al.*, 2001; Chipaux *et al.*, 2002). Most matrices studied are glasses and single-crystal materials; here we report a new ytterbium single-crystal matrix prepared using a KH_2PO_4 flux. This material belongs to a family of alkali rare-earth diphosphates with the general formula $M^I\text{LnP}_2\text{O}_7$ ($M^I = \text{Na, K, Rb, Cs, NH}_4$; $\text{Ln} = \text{rare-earth element}$). NaLnP_2O_7 ($\text{Ln} = \text{La, Ce, Eu, Yb}$) crystallizes in the orthorhombic or monoclinic systems (Horchani-Naifer & Férid, 2005; Férid & Horchani-Naifer, 2004; Férid, Horchani *et al.*, 2004; Férid, Horchani-Naifer *et al.*, 2004). KYbP_2O_7 is monoclinic ($P2_1/c$) and is isotypic with CsYbP_2O_7 (Jansen *et al.*, 1991). The structure consists of YbO_6 octahedra and P_2O_7 diphosphate groups connected to form a three-dimensional framework containing channels in which the K^+ ions are located (Fig. 1). The P_2O_7 group is formed from two slightly distorted PO_4 tetrahedra (Fig. 2).

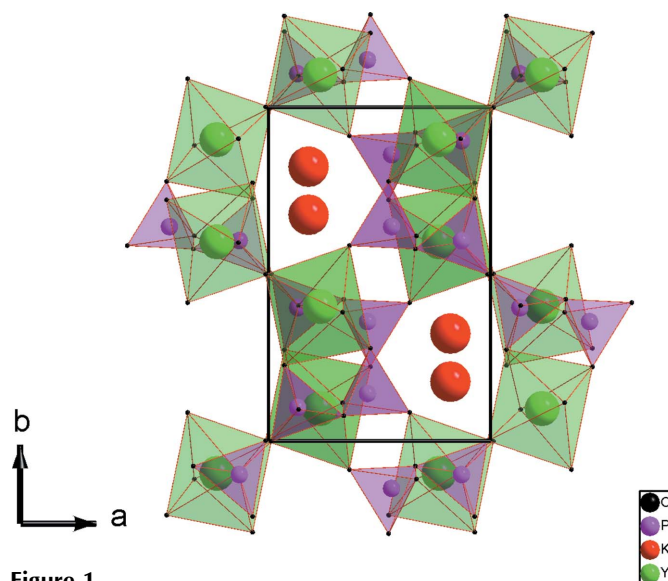


Figure 1
Projection of KYbP_2O_7 along the c -axis direction, showing the arrangement of YbO_6 octahedra and P_2O_7 groups, with K ions within channels.

In conformity with other diphosphates (Durif, 1995; Férid, 2006), the P—O distances range from 1.495 (5) to 1.612 (4) Å, the shortest corresponding to the terminal and the longest to the bridging O atoms.

Experimental

A mixture of KH₂PO₄ (10 g) and Yb₂O₃ (0.5 g) was gradually heated in a platinum crucible to 1273 K over 72 h, and then cooled and maintained at 873 K for 10 h. The crystals obtained were washed with hot water and hydrochloric acid.

Crystal data

KYb(P ₂ O ₇)	Z = 4
M _r = 386.08	D _x = 3.842 Mg m ⁻³
Monoclinic, P2 ₁ /c	Mo Kα radiation
a = 7.538 (3) Å	μ = 15.10 mm ⁻¹
b = 10.832 (2) Å	T = 298 (2) K
c = 8.537 (4) Å	Prism, colourless
β = 106.768 (7)°	0.18 × 0.17 × 0.17 mm
V = 667.4 (4) Å ³	

Data collection

Enraf–Nonius CAD-4 diffractometer	1957 independent reflections
ω/2θ scans	1763 reflections with I > 2σ(I)
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)	R _{int} = 0.044
T _{min} = 0.223, T _{max} = 0.243 (expected range = 0.070–0.077)	θ _{max} = 30.0°
10491 measured reflections	2 standard reflections every 150 reflections
	intensity decay: 2%

Refinement

Refinement on F ²	w = 1/[σ ² (F _o ²) + (0.0646P) ² + 0.1042P]
R[F ² > 2σ(F ²)] = 0.035	where P = (F _o ² + 2F _c ²)/3
wR(F ²) = 0.096	(Δ/σ) _{max} < 0.001
S = 1.17	Δρ _{max} = 4.91 e Å ⁻³
1957 reflections	Δρ _{min} = -3.27 e Å ⁻³
101 parameters	Extinction correction: SHELXL97
	Extinction coefficient: 0.0011 (1)

The highest peak and the deepest hole in the residual electron density are located 0.05 and 0.63 Å, respectively, from Yb.

Data collection: CAD-4 EXPRESS (Duisenberg, 1992; Macíček & Yordanov, 1992); cell refinement: CAD-4 EXPRESS; data reduction: XCAD4 (Harms & Wocadlo, 1995); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: DIAMOND (Brandenburg, 2001); software used to prepare material for publication: SHELXL97.

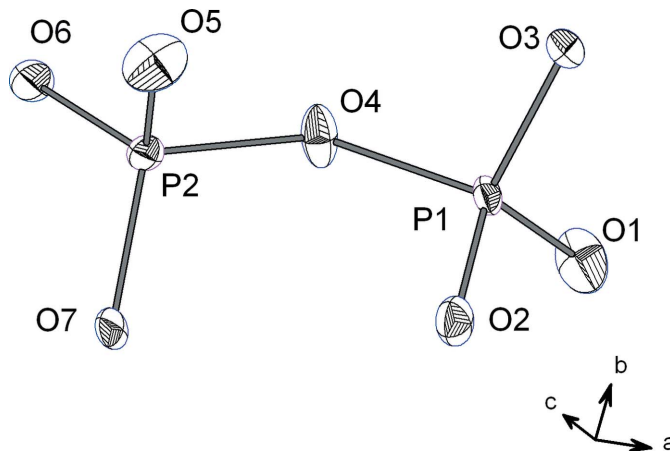


Figure 2
The P₂O₇ group, with displacement ellipsoids drawn at the 50% probability level.

References

Blasse, G. & Grabmaier, B. C. (1994). *Luminescent materials*. Berlin: Springer-Verlag.

Brandenburg, K. (2001). *DIAMOND*. Version 2.1e. Crystal Impact GbR, University of Bonn, Germany.

Chipaux, R., Cribier, M., Dujardin, C., Garnier, N., Guerassimova, N., Mallet, J., Meyer, J.-P., Pedrini, C. & Petrosyan, A. G. (2002). *Nucl. Instrum. Methods Phys. Res. Sect. A*, **486**, 228–233.

Duisenberg, A. J. M. (1992). *J. Appl. Cryst.* **25**, 92–96.

Durif, A. (1995). *Crystal Chemistry of Condensed Phosphates*. New York: Plenum Press.

Férid, M. (2006). *Etude des propriétés cristallographiques et physiques de phosphates condensés de terres rares*. Paris: Publibook.

Férid, M., Horchani, K. & Amami, J. (2004). *Mater. Res. Bull.* **39**, 1949–1995.

Férid, M. & Horchani-Naifer, K. (2004). *Mater. Res. Bull.* **39**, 2209–2217.

Férid, M., Horchani-Naifer, K. & Trabelsi-Ayadi, M. (2004). *Z. Kristallogr. New Cryst. Struct.* **219**, 353–354.

Guerassimova, N., Garnier, N., Dujardin, C., Petrosyan, A. G. & Pedrini, C. (2001). *Chem. Phys. Lett.* **339**, 197–202.

Harms, K. & Wocadlo, S. (1995). *XCAD4*. University of Marburg, Germany.

Hashimoto, N., Takada, Y., Sato, K. & Ibuki, S. (1991). *J. Lumin.* **48 & 49**, 893–897.

Horchani-Naifer, K. & Férid, M. (2005). *Solid State Ionics*, **176**, 1949–1953.

Horchani-Naifer, K., Férid, M., Gàcon, J. C. & Trabelsi-Ayadi, M. (2006). Tunisian patent SN 06 088.

Jansen, M., Wu, G. Q. & Königstein, K. Z. (1991). *Z. Kristallogr.* **197**, 245–246.

Macíček, J. & Yordanov, A. (1992). *J. Appl. Cryst.* **25**, 73–80.

Pieteron, L. van, Heeroma, M., de Heer, E. & Meijerink, A. (2000). *J. Lumin.* **91**, 177–193.

Sheldrick, G. M. (1996). *SADABS*. University of Göttingen, Germany.

Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.

Weber, M. J. (1999). *Handbook of Laser Wavelengths*. Boca Raton: CRC Press.